# JEE-Main-28-07-2022-Shift-2 (Memory Based)

# **Physics**

**Question:** Two photons, first has energy twice the other. If energy of first photon is 5 times its work function, find the ratio of maximum velocities

**Options:** 

- (a)  $\frac{3}{2}$
- (b)  $\frac{2}{3}$
- (c)  $\frac{4}{3}$
- (d)  $\frac{4}{9}$

Answer: (b)

Solution:

Max. K.E. of electron

$$= E - \phi$$

$$\frac{1}{2}mv_1^2 = E_1 - \phi$$

$$\frac{1}{2}mv_2^2 = E_2 - \phi$$

$$\frac{{v_1}^2}{{v_2}^2} = \frac{5\phi - \phi}{10\phi - \phi} = \frac{4}{9}$$

$$\frac{v_1}{v_2} = \sqrt{\frac{4}{9}} = \frac{2}{3}$$

**Question:** A capacitor has capacity  $C_0$ . When a dielectric of constant K& width  $3/4^{th}$  of separation between plates is inserted, then find the new capacity.

Options:

(a) 
$$\left(\frac{7k}{K+3}\right)C_0$$

(b) 
$$\left(\frac{5k}{K+3}\right)C_0$$

(c) 
$$\left(\frac{5k}{K+3}\right)C_0$$

(d) 
$$\left(\frac{4k}{K+3}\right)C_0$$

$$C = \frac{\varepsilon_0 A}{\left(d - t + \frac{t}{k}\right)} \text{ and } C_0 = \frac{\varepsilon_0 A}{d}$$

Here 
$$t = \frac{3}{4}d$$

$$\therefore C = \frac{\varepsilon_0 A}{\left(d - \frac{3}{4}d + \frac{3d}{4k}\right)} = \frac{\varepsilon_0 A}{\frac{d}{4k}(3+k)}$$

$$=\frac{4k}{\left(k+3\right)}C_0$$

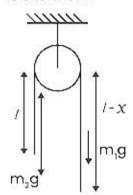
Question: A chain of uniform mass desity with length 'L', mass M is hanging over an ideal smooth pulley. When  $l = \frac{L}{x}$ , the acceleration of chain is  $\frac{g}{2}$ , find x?

## **Options:**

- (d)  $\frac{L}{1}$

# Answer: (a)

#### Solution:



$$m_1 = \frac{M}{L} (L - x) g$$

$$m_2 = \frac{M}{L} xg$$

$$m_{2} = \frac{M}{L}xg$$

$$a = \frac{m_{1}g - m_{2}g}{M} = \frac{g}{2} \frac{\frac{M}{2}(L - x)g - \frac{M}{2}xg}{M} = \frac{g}{2}$$

$$\frac{L - x - x}{L} = \frac{1}{2}$$

$$L - 2x = \frac{L}{2}$$

$$2x = L \frac{L}{2} = \frac{L}{2} \Rightarrow x = \frac{L}{4}$$

**Question:** Time period of a pendulum is measured on surface of a planet with a stop watch of 1 sec isolation is found to be 200 sec. Its length was measured as  $1\pm0.001m$ . percentage error in measurement of arc due to gravity is

#### **Options:**

- (a) 1.1%
- (b) 0.3%
- (c) 2.5%
- (d) 5.2%

Answer: (a)

Solution:

as 
$$T = 2\pi \sqrt{\frac{l}{g}}$$
  $\therefore \Delta g\% = 2 \times \frac{1}{200} \times 100 + \frac{0.001}{1} \times 100$   
or  $\frac{\Delta g}{g} = \frac{2\Delta T}{T} + \frac{\Delta l}{l}$  = 1 + 0.1  
= 1.1%

**Question:** Water flows out of a pipe and hits a wall, it has horizontal velocity v and cross-sectional area A. The density of the water is p. The water does not rebound from the wall. What is the fore exerted on the wall by the water?

#### **Options:**

- (a)  $F = \rho v^2 A$
- (b)  $F = \rho v^{1} A$
- (c)  $F = \rho v A$
- (d)  $F = \rho v^2$

Answer: (a)

Solution:

$$F = \frac{d}{dt} \left[ \vec{p}_{\text{of water}} \right] = \left( \frac{dm}{dt} \right) v$$
$$= \rho V A. v$$
$$F = \rho v^2 A$$

**Question:** Time period of a pendulum is measured on surface of a planet with a stop watch of 1 sec resolution in found to be 200 sec. Its length was measured as  $1 \pm 0.001m$ . Percentage error in measurement of are due to gravity is

- (a) 1.4%
- (b) 1.1%
- (c) 2.2%
- (d) 2.1%

Answer: (b)

#### Solution:

$$T = 2\pi \sqrt{\frac{l}{g}}$$

$$\Rightarrow g = \frac{T^2}{4\pi^2 l}$$

$$\frac{\Delta g}{g} = 2\frac{\Delta T}{T} + \frac{\Delta l}{l}$$

$$= 2\left(\frac{1}{200}\right) + \frac{0.001}{1}$$

$$= \frac{1}{100} + 0.001 = 0.011$$

% error = 1.1%

**Question:** Ball thrown upwards from a building with speed 19.6 m/s. Find maximum height achieved from point of projection?  $(g = 9.8m / s^2)$ 

# **Options:**

- (a) 1.96 m
- (b) 2.9 m
- (c) 19.6 m
- (d) 12.96 m

Answer: (c)

#### Solution:

Maximum height = h (say)

$$v^2 - u^2 = 2gh$$

At topmost point v = 0

And u = 19.6m / s (given)

$$0^2 - (19.6)^2 = -2(9.8)(h)$$

$$h = \frac{\left(19.6\right)^2}{19.6} = 19.6m$$

**Question:** Time period of a simple pendulum on earth's surface is 4 sec and at height h above the earth's surface is 6 sec. Find the value of h if R = 6400 km.

### **Options:**

- (a) 1200
- (b) 2400

Answer: (c)

Solution:

$$T = 2\pi \sqrt{\frac{l}{g}}$$

$$\therefore \frac{T_1}{T_2} = \sqrt{\frac{g_2}{g_1}} = \sqrt{\frac{\frac{GM}{(R+h)^2}}{\frac{GM}{R^2}}}$$

$$\frac{4}{6} = \frac{R}{R+h}$$

$$\Rightarrow 4R + 4h = 6R \Rightarrow 4h = 2R$$

$$h = \frac{R}{2} = 3200$$

**Question:** For the given equation, please select incorrect statement, where K is Boltzmann constant, T is absolute temperature.

$$\sin \theta = \alpha \beta \ln \left[ \frac{Bx}{KT} \right]$$

#### **Options:**

- (a) Dimension of  $\beta$  is same as force
- (b) Dimension of  $\alpha^{-1}x$  is same as energy
- (c) Dimension of  $\alpha$  and  $\beta$  is same
- (d) Dimension of KT is same as energy

Answer: (c)

Solution:

$$[B] = \frac{[KT]}{[x]} = \frac{M^1 L^2 T^{-2}}{L} = [M^1 L^1 T^{-2}]$$

$$\alpha = \frac{M^{0}L^{0}T^{0}}{\beta} = [M^{-1}L^{-1}T^{+2}]$$

**Question:** Assertion: Resistance of  $80\Omega$  is cut equally in 4 parts and all resistances are kept parallelly the net resistance is  $5\Omega$ 

**Reason:** When 2R and 3R connected in parallel, ratio of heat dissipated in them is 3:2 correct statement is/are

#### **Options:**

- (a) If both assertion and reason are true and the reason is the correct explanation of the assertion.
- (b) If both assertion and reason are true, but the reason is not the correct explanation of the assertion.
- (c) If assertion is true, but reason is false.
- (d) If both the assertion and reason are true.

Assertion: Let r be resistance of each cut part then  $r = \frac{80}{4} = 20\Omega$ 

Now 4 resistors of 20 $\Omega$  in parallel  $R_{eff} = \frac{20}{4} = 5\Omega$ 

Reason: 
$$\frac{H_1}{H_2} = \frac{\frac{V^2}{R_1}}{\frac{V^2}{R_2}} = \frac{R_2}{R_1} = \frac{3}{2}$$

**Question:** Two light rays incident on surface of a metal with energies 5 times and 6 times the work function f metal respectively. The ratio of speeds of electrons, ejected with maximum kinetic energies in each case, respectively is

## **Options:**

- (a)  $\frac{4}{5}$
- (b)  $\frac{2}{5}$
- (c)  $\frac{4}{\sqrt{5}}$
- (d)  $\frac{2}{\sqrt{5}}$

# Answer: (d)

#### Solution:

As 
$$KE_{\text{max}} = \frac{1}{2}mv^2 = hf - W$$

$$\therefore \frac{1}{2}mv_1^2 = 5W - W = 4W$$

And 
$$\frac{1}{2}mv_2^2 = 6W - W = 5W$$

$$\left(\frac{V_1}{V_2}\right)^2 = \frac{4}{5}$$

Or 
$$\frac{V_1}{V_2} = \frac{2}{\sqrt{5}}$$

**Question:** Work done by gas in expansion is 150J and degrees of freedom = 8 find heat (q) **Options:** 

- (a) 250
- (b) 350
- (c) 650
- (d) 750

$$\omega = \rho \Delta V = nR\Delta T = 150J \dots (1)$$

$$Q = nC_p \Delta T$$

$$= n \left(\frac{f+2}{2}\right) R \Delta T$$

$$= \left(\frac{8+2}{2}\right) 150 = 5 \times 150 = 750$$

Question: Work done by gas in isobaric expansion is 150J and degrees of freedom = 8. Find heat (q)

# **Options:**

- (a) 750 J
- (b) 650 J
- (c) 550 J
- (d) 450 J

Answer: (a)

Solution:

$$\Delta U = \frac{nf}{2} R \Delta T = 4nR \Delta T$$

$$\Delta W = 150J = nR\Delta T$$

$$\Delta Q = \Delta W + \Delta U$$

$$= nR\Delta T + 4nR\Delta T$$

$$=5nR\Delta T$$

$$=5(150) = 750J$$

Question: The ratio of magnetic field at the centre of circular coil and at a distance of  $\sqrt{3}R$ from centre on axis of the coil is

# **Options:**

- (a) 1:1
- (b) 4:2
- (c) 7:3
- (d) 8:1

Answer: (d)

Solution:

$$B_{centre} = \frac{\mu_0 i}{2R}$$

$$B_{centre} = \frac{\mu_0 i}{2R}$$

$$B_{axis} = \frac{\mu_0 i R^2}{2(R^2 + x^2)^{3/2}}$$

$$= \frac{\mu_0 i R^2}{2 \left(3R^2 + R^2\right)^{3/2}} = \frac{\mu_0 i}{16R}$$

$$\Rightarrow \frac{B_{centre}}{B_{axis}} = \frac{8}{1}$$

# JEE-Main-28-07-2022-Shift-2 (Memory Based)

# **Chemistry**

Question: Conc. HNO3 reacts with I2 to give

#### **Options:**

- (a) HI
- (b) HOI
- (c) HIO<sub>3</sub>
- (d) HIO<sub>2</sub>

Answer: (c)

Solution: Iodine reacts with concentrated nitric acid according to the following equation

 $10HNO_3 + I_2 \rightarrow 2HIO_3 + 10NO_2 + 4H_2O$ 

Question: White phosphorus reacts with thionyl chloride to give

#### **Options:**

- (a) POCl<sub>3</sub>
- (b) PCl<sub>3</sub>
- (c) SO<sub>2</sub>Cl<sub>2</sub>
- (d) SCl<sub>3</sub>

Answer: (c)

**Solution:**  $P_4 + 8SOCl_2 \rightarrow 4PCl_3 + 4SO_2 + 2S_2Cl_2$ 

Question: Products obtained during treatment of hard water using Clark's method are

# **Options:**

- (a) CaCO<sub>3</sub>, Mg(OH)<sub>2</sub>
- (b) Ca(OH)2, MgCO3
- (c) CaCO<sub>3</sub>, MgCO<sub>3</sub>
- (d) Ca(OH)2, Mg(OH)2

Answer: (a)

**Solution:** In Clark's method calculated amount of lime is added to hard water. It precipitates out calcium carbonate and magnesium hydroxide which can be filtered off.

$$Ca(HCO_3)_2 + Ca(OH)_2 \rightarrow 2CaCO_3 \downarrow + 2H_2O$$

$$Mg(HCO_3)_2 + 2Ca(OH)_2 \rightarrow 2CaCO_3 \downarrow + Mg(OH)_2 \downarrow + 2H_2O$$

Question: Decreasing order of metallic character

Na, Mg, Be, Si, P

#### **Options:**

- (a) Na > Mg > Be > P > Si
- (b) Mg > Na > Si > Be > P
- (c) Na > Mg > Be > Si > P
- (d) P > Si > Be > Mg > Na

Answer: (c)

**Solution:** Metallic character increases down the group and decreases along the period as we move left to right.

Hence, order of decreasing metalic character is Na > Mg > Be > Si > P

Question: Assertion: Permanganate titrations are not performed in presence of HCl

Reason: Chlorine is formed as a consequence of oxidation of HCl

#### **Options:**

- (a) Both assertion and reason are true, reason is correct explanation of assertion.
- (b) Both assertion and reason are true, but reason is not a correct explanation of the assertion.
- (c) Assertion is true, but reason is false
- (d) Assertion is false, but reason is true

Answer: (a)

**Solution:** Permanganate titrations are not performed in presence of HCl because HCl oxidises to form chlorine.

#### Question:

#### **Options:**

(a)

(b)

(c)

(d)

Answer: (a) Solution:

#### Question: Match the following.

(Column I)	(Column II)
Compounds	Hybridizations
(A) Ni(CO) <sub>4</sub>	$(i) d^2sp^3$
(B) [Ni(CN) <sub>4</sub> ] <sup>2-</sup>	(ii) sp <sup>3</sup>
(C) [Co(CN) <sub>6</sub> ] <sup>4-</sup>	(iii) dsp <sup>2</sup>

#### **Options:**

- (a)  $A \rightarrow (ii)$ ;  $B \rightarrow (iii)$ ;  $C \rightarrow (i)$
- (b)  $A \rightarrow (iii)$ ;  $B \rightarrow (i)$ ;  $C \rightarrow (ii)$
- (c)  $A \rightarrow (iii)$ ;  $B \rightarrow (ii)$ ;  $C \rightarrow (i)$
- (d)  $A \rightarrow (i)$ ;  $B \rightarrow (iii)$ ;  $C \rightarrow (ii)$

Answer: (a)

# Solution:

- A)  $Ni(CO)_4 \Rightarrow sp^3$
- B)  $[Ni(CN)_4]^{2-} \Rightarrow dsp^2$
- C)  $[Co(CN)_6]^{4-} \Rightarrow d^2sp^3$

Question: Assertion: Aniline on nitration gives o, p, m products

Reason: Mixture used in nitration is acidic

## **Options:**

- (a) Both assertion and reason are true, reason is correct explanation of assertion.
- (b) Both assertion and reason are true, but reason is not a correct explanation of the assertion.
- (c) Assertion is true, but reason is false
- (d) Assertion is false, but reason is true

Answer: (a)

**Solution:** Direct nitration of aniline yields tarry oxidation products in addition to the nitro derivatives. Moreover, in the strongly acidic medium, aniline is protonated to form the anilinium ion which is meta directing. That is why besides the ortho and para derivatives, significant amount of meta derivative is also formed.

Question: Assertion: Zero orbital overlap is an out of phase overlap

**Reason:** It results due to different orientation / direction of approach of orbitals **Options:** 

- (a) Both assertion and reason are true, reason is correct explanation of assertion.
- (b) Both assertion and reason are true, but reason is not a correct explanation of the assertion.
- (c) Assertion is true, but reason is false
- (d) Assertion is false, but reason is true

Answer: (a)

Solution: Zero overlap is out of phase due to different orientation direction of approach

**Question:** Find the change in oxidation state of oxygen when KIO<sub>4</sub> reacts with H<sub>2</sub>O<sub>2</sub>? **Options:** 

- (a) 4
- (b) 2
- (c) 1
- (d)3

Answer: (c)

Solution:  $KIO_4 + H_2 \overset{-1}{O} \rightarrow KIO_3 + H_2O + \overset{0}{O}_2$ 

Question:

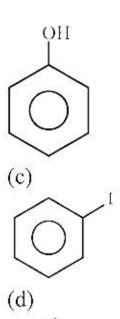
$$\begin{array}{c|c}
 & HI \\
\hline
 & Heat
\end{array}$$
A Zn dust
B

**Options:** 

(a)



(b)



Answer: (a) Solution:

Question: Matching of polymers:

(Column I)	(Column II)	
(A) Nylon	(i) Thermosetting	
(B) Bakelite	(ii) Thermoplastic	
(C) Polythene	(iii) Elastomer	
(D) Natural rubber	(iv) Fibers	

#### **Options:**

(a) 
$$A \rightarrow (i)$$
;  $B \rightarrow (iii)$ ;  $C \rightarrow (ii)$ ;  $D \rightarrow (iv)$ 

(b) A 
$$\rightarrow$$
 (iv); B  $\rightarrow$  (i); C  $\rightarrow$  (ii); D  $\rightarrow$  (iii)

(c) A 
$$\rightarrow$$
 (iii); B  $\rightarrow$  (ii); C  $\rightarrow$  (iv); D  $\rightarrow$  (i)

(d) A 
$$\rightarrow$$
 (ii); B  $\rightarrow$  (i); C  $\rightarrow$  (iv); D  $\rightarrow$  (iii)

Answer: (b)

# Solution:

Nylon ⇒ Fibers

 $Bakelite \Rightarrow Thermosetting$ 

Polythene  $\Rightarrow$  Thermoplastic

Natural rubber ⇒ Elastomer

Question: Arrange the following according to the rate of nitration:

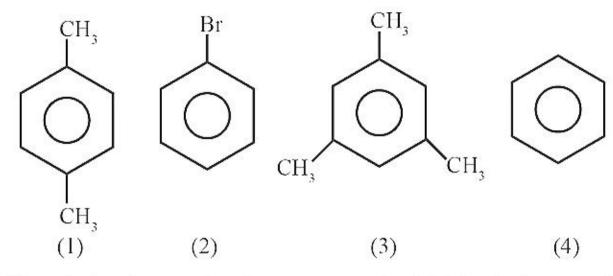
1) p - xylene

- 2) bromobenzene
- 3) mesitylene
- 4) benzene

- (a) 3 > 1 > 4 > 2
- (b) 2 > 1 > 4 > 3
- (c) 1 > 3 > 2 > 4
- (d) 4 > 2 > 3 > 1

Answer: (a)

#### Solution:



More is the electron density groups attached, higher is the rate of nitration.

**Question:** Find the molarity of Na<sub>2</sub>SO<sub>4</sub> formed upon mixing of 0.2 M, 2 L H<sub>2</sub>SO<sub>4</sub> and 0.1 M, 2 L NaOH.

#### **Options:**

- (a) 0.05 M
- (b) 0.03 M
- (c) 0.04 M
- (d) 0.025 M

Answer: (d)

#### Solution:

$$H_2SO_4 + 2NaOH \rightarrow Na_2SO_4 + 2H_2O$$

Initial moles

0.4

0.2

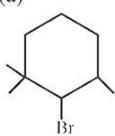
. .

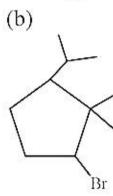
NaOH is the limiting reagent

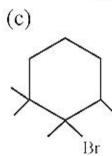
- $\therefore$  2 moles of NaOH  $\rightarrow$  1 mole of Na<sub>2</sub>SO<sub>4</sub>
- ∴ 0.2 moles of NaOH  $\rightarrow \frac{0.2}{2} = 0.1$  moles of Na<sub>2</sub>SO<sub>4</sub>

$$M_{\text{Na}_2\text{SO}_4} = \frac{0.1}{4} = 0.025 \,\text{M}$$

# Question:







(d)

Answer: (c) Solution:

**Question:** The temperature at which rms speed of gas molecules becomes double of its value at 0°C is

## **Options:**

- (a) 819°C
- (b) 760°C
- (c) 273°C
- (d) 224°C

#### Answer: (a)

Solution:

$$\begin{split} V_{rms} &= \sqrt{\frac{3RT}{M}} \\ V_{rms} &\propto \sqrt{T} \\ \frac{\left(V_{rms}\right)_{l}}{\left(V_{rms}\right)_{2}} &= \sqrt{\frac{T_{l}}{T_{2}}} \\ \left(V_{rms}\right)_{2} &= 2\left(V_{rms}\right)_{l} \\ \frac{1}{2} &= \sqrt{\frac{273}{T_{2}}} \\ \Rightarrow T_{2} &= 273 \times 4 = 1092 \text{ K} \\ &= 819^{\circ}\text{C} \end{split}$$

**Question:** In the reaction,  $N_2 + O_2 \rightleftharpoons 2NO$ 

If initial moles of  $N_2$  and  $O_2$  are 2 mol and 1 mol, then find the value of  $K_c$ . The moles of  $O_2$  at equilibrium is 0.6 mol.

Answer: 0.67

Initial moles 2 1 -  
Final moles 2-0.4 1-0.4 0.4 × 2  
= 1.6 = 0.6 = 0.8  

$$K_{C} = \frac{[NO]^{2}}{[N_{2}][O_{2}]} = \frac{\left(\frac{0.8}{V}\right)^{2}}{\left(\frac{1.6}{V}\right)\left(\frac{0.6}{V}\right)} = 0.67$$

**Question:** In a FCC lattice a molecule, density = 9.03 g/cm<sup>3</sup>, a =  $4 \times 10^{-8}$  cm. Find molecular weight

**Answer:** 87.00

**Solution:** Z = 4, density = 9.03 cm<sup>-3</sup>,  $a = 4 \times 10^{-8}$  cm

$$d = \frac{ZM}{a^3 N_a}$$

$$M = \frac{da^{3}N_{a}}{Z} = \frac{9.03 \times (4 \times 10^{-8})^{3} \times 6.02 \times 10^{23}}{4} = 869.7 \times 10^{-1}$$

= 86.97 g/mol

 $M \approx 87 \text{ g/mol}$ 

Question: How many of the following are not isoelectronic species?

- A) Yb<sup>-3</sup>
- B) Tb<sup>+4</sup>
- C) Eu<sup>+3</sup>
- D) Tb<sup>+2</sup>

Answer: 4.00

Solution:

 $Yb^{+3} - 4f^{13}$  (67 electrons)

 $Tb^{+4} - 4f^{7}$  (61 electrons)

 $Eu^{+3} - 4f^6$  (60 electrons)

 $Tb^{+2} - 4f^{9}$  (63 electrons)

None of them is isoelectronic species

# JEE-Main-28-07-2022-Shift-2 (Memory Based)

# **MATHEMATICS**

Question: If 
$$P\left(\frac{B}{A}\right) = \frac{5}{7}$$
;  $P\left(\frac{A}{B}\right) = \frac{7}{9}$  and  $P(A \cap B) = \frac{1}{9}$ . Given

$$S_1 \equiv P(A' \cup B) = \frac{5}{6}, S_2 \equiv P(A' \cap B') = \frac{1}{18}$$
, then:

#### **Options:**

- (a) Both  $S_1$  and  $S_2$  are correct
- (b)  $S_1$  is true and  $S_2$  is false
- (c)  $S_1$  is false and  $S_2$  is true
- (d) Both  $S_1$  and  $S_2$  are false

#### Answer: (d)

#### Solution:

Given: 
$$\frac{P(A \cap B)}{P(A)} = \frac{5}{7} & \frac{P(A \cap B)}{P(B)} = \frac{7}{9}$$

As 
$$P(A \cap B) = \frac{1}{9}$$
, we get

$$P(A) = \frac{7}{45} & P(B) = \frac{1}{7}$$

$$P(A' \cap B') = 1 - P(A \cap B)$$

$$=1 - \left(\frac{7}{45} + \frac{1}{7} - \frac{1}{9}\right)$$

$$=\frac{256}{315}$$

Given, 
$$S_1 \equiv P(A' \cup B) = \frac{5}{6}$$

$$S_2 \equiv P(A' \cap B') = \frac{1}{18}$$
, then:

As 
$$P(A \cap B) = \frac{1}{9}$$
, we get  $P(A) = \frac{7}{45} \& P(B) = \frac{1}{7}$ 

$$P(A' \cup B) = 1 - (P(A) - P(A \cap B))$$

$$=1-\left(\frac{7}{45}-\frac{1}{9}\right)=\frac{43}{45}$$

 $\therefore S_1$  is wrong.

**Question:** Absolute maximum value of  $f(x) = \tan^{-1}(\sin x - \cos x)$  is:

- (a) 0
- (b)  $\tan^{-1} \frac{1}{\sqrt{2}} \frac{\pi}{4}$
- (c)  $\frac{\pi}{4}$
- (d)  $\tan^{-1} \sqrt{2}$

Answer: (d)

Solution:

$$\because \sin x - \cos x \in \left[ -\sqrt{2}, \sqrt{2} \right]$$

So maximum value of  $\tan^{-1} (\sin x - \cos x)$  is  $\tan^{-1} \sqrt{2}$ 

Question: The values of  $\lambda$  for which the lines  $\frac{x-1}{1} = \frac{y-2}{2} = \frac{z+3}{\lambda^2}$  and  $\frac{x-3}{1} = \frac{y-2}{\lambda^2} = \frac{z-1}{2}$  are coplanar, are:

**Options:** 

- (a)  $\pm \sqrt{3}$
- (b)  $\pm \sqrt{5}$
- $(c) \pm 2$
- (d)  $\pm \sqrt{2}$

Answer: (d)

Solution:

For coplanarity

$$\begin{vmatrix} 1 & 2 & \lambda^2 \\ 1 & \lambda^2 & 2 \\ 2 & 0 & 4 \end{vmatrix} = 0$$

$$\Rightarrow 1(4\lambda^2) - 2(0) + \lambda^2(-2\lambda^2) = 0$$

$$\Rightarrow -2\lambda^4 + 4\lambda^2 = 0$$

$$\Rightarrow \lambda^2 \left( \lambda^2 - 2 \right) = 0$$

$$\Rightarrow \lambda = \pm \sqrt{2}$$

**Question:** From point (2,0) tangents are drawn on  $2y^2 = -x$ . These tangents also touches the circle  $(x-5)^2 + y^2 = r^2$ . The value of  $17r^2$  is:

**Options:** 

- (a) 1
- (b) 12
- (c) 9
- (d) 4

Answer: (c)

$$P \equiv y^2 = \frac{-x}{2}$$

Equation of tangent  $\equiv y = mx + \left(-\frac{1}{8m}\right)$ 

$$16m^2 = 1 \Longrightarrow m = \pm \frac{1}{4}$$

$$\therefore$$
 Tangent are  $y = \frac{1}{4}x - \frac{1}{2} \& y = -\frac{x}{4} + \frac{1}{2}$ 

Equation of tangent to  $(x-5)^2 + y^2 = r^2$  are:

$$x - 4y = 2 & 4y + x = 2$$

Using  $d_c = r$ , we get

$$\left| \frac{5 - 0 - 2}{\sqrt{17}} \right| = r \text{ or } \left| \frac{5 + 0 - 2}{\sqrt{17}} \right| = r$$

$$\therefore r^2 = \frac{9}{17}$$

$$\Rightarrow 17r^2 = 9$$

Question: Let  $f(x) = \lim_{n \to \infty} \frac{\cos 2\pi x - x^{2n} \sin(x-1)}{1 + x^{2n+1} - x^{2n}}$ , is continuous at:

## **Options:**

(a) 
$$R - \{1\}$$

(b) 
$$R - \{-1,1\}$$

(c) 
$$R - \{0,1\}$$

(d) 
$$R - \{0\}$$

# Answer: (b)

#### Solution:

For 
$$|x| < 1$$
,  $f(x) = \cos 2\pi x$ 

For 
$$|x| > 1$$
,  $f(x) = -\frac{\sin(x-1)}{x-1}$ 

For 
$$|x| = 1$$
,  $f(x) = \begin{cases} 1 & \text{if } x = 1\\ \frac{(1+\sin 2)}{-1} & \text{if } x = -1 \end{cases}$ 

$$\lim_{x \to 1^{-}} f(x) = -1, \lim_{x \to 1^{-}} f(x) = 1$$

So f is discontinuous at x = 1

$$\lim_{x \to -1^{+}} f(x) = 1, \lim_{x \to -1^{-}} f(x) = -\frac{\sin 2}{2}$$

So f(x) is discontinuous at x = -1

**Question:** A class have B boys and G girls, 3 boys and 2 girls selected at random and number of ways of selecting 3 boys and 2 girls are 168. Then B+3G is equal to \_\_\_\_\_.

**Answer: 17.00** 

#### Solution:

Given that

$$\begin{array}{l}
{}^{B}C_{3} \cdot {}^{G}C_{3} = 168 \\
\Rightarrow \frac{B(B-1)(B-2)}{6} \cdot \frac{G(G-1)}{2} = 168 \\
\Rightarrow B(B-1)(B-2)G(G-1) = 7 \cdot 6 \cdot 4 \cdot 3 \cdot 2 \cdot 2 \\
\Rightarrow B(B-1)(B-2)G(G-1) = 8 \cdot 7 \cdot 6 \cdot 3 \cdot 2 \\
\therefore B = 8 & G = 3
\end{array}$$

**Question:** Let  $f(x) = ax^2 + bx + c$  and f(1) = 3,  $f(-2) = \lambda$ , f(3) = 4, then the value of  $\lambda$  for which f(0) + f(1) + f(-2) + f(3) = 14 is \_\_\_\_.

Answer: 4.00

#### Solution:

Given, 
$$f(x) = ax^2 + bx + c$$

 $\Rightarrow B + 3G = 8 + 9 = 17$ 

$$f(0) = c$$

$$f(1) = a + b + c = 3$$
 ....(1)

$$f(-2) = 4a - 2b + c = \lambda$$
 ....(2)

$$f(3) = 9a + 3b + c = 4$$
 ....(3)

By solving (2) & (3)

$$a+b=\frac{4-\lambda}{5}$$

Also, 
$$c = 3 + \lambda + 4 = 14$$

$$c = 7 - \lambda$$

$$\frac{4-\lambda}{5} + 7 - \lambda = 3$$

$$6\lambda = 24$$

$$\lambda = 4$$