

Q:1
Topic Name:Mathematics-Section A

ItemCode:191

The number of points of intersection of $|z - (4 + 3i)| = 2$ and $|z| + |z - 4| = 6$,

Question: $z \in \mathbb{C}$, is

A	0
B	1
C	2
D	3

Q:2
Topic Name:Mathematics-Section A

ItemCode:192

Let $f(x) = \begin{vmatrix} a & -1 & 0 \\ ax & a & -1 \\ ax^2 & ax & a \end{vmatrix}$, $a \in \mathbb{R}$. Then the sum of the squares of all the values

Question: of a , for which $2f'(10) - f'(5) + 100 = 0$, is

A	117
B	106
C	125
D	136

Q:3
Topic Name:Mathematics-Section A

ItemCode:193

Let for some real numbers α and β , $\alpha = \alpha - i\beta$. If the system of equations $4ix + (1 + i)y = 0$ and $8\left(\cos\frac{2\pi}{3} + i\sin\frac{2\pi}{3}\right)x + \overline{\alpha}y = 0$ has more than one solution,

Question: then $\frac{\alpha}{\beta}$ is equal to

A	$-2 + \sqrt{3}$
B	$2 - \sqrt{3}$
C	$2 + \sqrt{3}$
D	$-2 - \sqrt{3}$

Q:4
Topic Name:Mathematics-Section A

ItemCode:194

Let A and B be two 3×3 matrices such that $AB = I$ and $|A| = \frac{1}{8}$. Then

Question: $|\text{adj}(B \text{adj}(2A))|$ is equal to

A	16
B	32
C	64
D	128

Q:5

ItemCode:195

Question: Let $S = 2 + \frac{6}{7} + \frac{12}{7^2} + \frac{20}{7^3} + \frac{30}{7^4} + \dots$. Then $4S$ is equal to

A $\left(\frac{7}{3}\right)^2$

B $\frac{7^3}{3^2}$

C $\left(\frac{7}{3}\right)^3$

D $\frac{7^2}{3^3}$

Q:6

Topic Name: Mathematics-Section A

ItemCode:196

Question: If $a_1, a_2, a_3, \dots$ and $b_1, b_2, b_3, \dots$ are A.P., and $a_1 = 2, a_{10} = 3, a_1 b_1 = 1 = a_{10} b_{10}$, then $a_4 b_4$ is equal to -

A $\frac{35}{27}$

B 1

C $\frac{27}{28}$

D $\frac{28}{27}$

Q:7

Topic Name: Mathematics-Section A

ItemCode:197

If m and n respectively are the number of local maximum and local minimum

points of the function $f(x) = \int_0^{x^2} \frac{t^2 - 5t + 4}{2 + e^t} dt$, then the ordered pair (m, n) is

Question: equal to

A (3, 2)

B (2, 3)

C (2, 2)

D (3, 4)

Q:8

Topic Name: Mathematics-Section A

ItemCode:198

Let f be a differentiable function in $\left(0, \frac{\pi}{2}\right)$. If $\int_{\cos x}^1 t^2 f(t) dt = \sin^3 x + \cos x$,

then $\frac{1}{\sqrt{3}} f'\left(\frac{1}{\sqrt{3}}\right)$ is equal to

Question:

A $6 - 9\sqrt{2}$

B $6 - \frac{9}{\sqrt{2}}$

C $\frac{9}{2} - 6\sqrt{2}$

D $\frac{9}{\sqrt{2}} - 6$

Q:9

Topic Name:Mathematics-Section A

ItemCode:199	
The integral $\int_0^1 \frac{1}{7^{\left[\frac{1}{x}\right]}} dx$, where $[\cdot]$ denotes the greatest integer function, is equal to	
Question:	
A	$1 + 6 \log_{\varepsilon} \left(\frac{6}{7} \right)$
B	$1 - 6 \log_{\varepsilon} \left(\frac{6}{7} \right)$
C	$\log_{\varepsilon} \left(\frac{7}{6} \right)$
D	$1 - 7 \log_{\varepsilon} \left(\frac{6}{7} \right)$

Q:10

Topic Name:Mathematics-Section A

ItemCode:1910	
If the solution curve of the differential equation $((\tan^{-1} y) - x) dy = (1 + y^2) dx$ passes through the point $(1, 0)$, then the abscissa of the point on the curve whose	
Question: ordinate is $\tan(1)$, is	
A	$2e$
B	$\frac{2}{e}$
C	2
D	$\frac{1}{e}$

Q:11

Topic Name:Mathematics-Section A

ItemCode:1911	
If the equation of the parabola, whose vertex is at $(5, 4)$ and the directrix is $3x + y - 29 = 0$, is $x^2 + ay^2 + bxy + cx + dy + k = 0$, then $a + b + c + d + k$ is equal	
Question: to	
A	575
B	-575
C	576
D	-576

Q:12

Topic Name:Mathematics-Section A

ItemCode:1912	
The set of values of k , for which the circle $C : 4x^2 + 4y^2 - 12x + 8y + k = 0$ lies	
Question: inside the fourth quadrant and the point $\left(1, -\frac{1}{3}\right)$ lies on or inside the circle C , is	
A	an empty set
B	$\left(6, \frac{65}{9}\right]$
C	$\left[\frac{80}{9}, 10\right)$
D	$\left(9, \frac{92}{9}\right]$

Q:13

Topic Name:Mathematics-Section A

ItemCode:1913

Let the foot of the perpendicular from the point $(1, 2, 4)$ on the line

$\frac{x+2}{4} = \frac{y-1}{2} = \frac{z+1}{3}$ be P . Then the distance of P from the plane

Question: $3x + 4y + 12z + 23 = 0$ is

- A 5
- B $\frac{50}{13}$
- C 4
- D $\frac{63}{13}$

Q:14

Topic Name:Mathematics-Section A

ItemCode:1914

The shortest distance between the lines $\frac{x-3}{2} = \frac{y-2}{3} = \frac{z-1}{-1}$ and

$\frac{x+3}{2} = \frac{y-6}{1} = \frac{z-5}{3}$, is

Question:

- A $\frac{18}{\sqrt{5}}$
- B $\frac{22}{3\sqrt{5}}$
- C $\frac{46}{3\sqrt{5}}$
- D $6\sqrt{3}$

Q:15

Topic Name:Mathematics-Section A

ItemCode:1915

Let $\vec{a}$ and $\vec{b}$ be the vectors along the diagonals of a parallelogram having area

$2\sqrt{2}$. Let the angle between $\vec{a}$ and $\vec{b}$ be acute, $|\vec{a}| = 1$, and $|\vec{a} \cdot \vec{b}| = |\vec{a} \times \vec{b}|$.

If $\vec{c} = 2\sqrt{2}(\vec{a} \times \vec{b}) - 2\vec{b}$, then an angle between $\vec{b}$ and $\vec{c}$ is

Question:

- A $\frac{\pi}{4}$
- B $-\frac{\pi}{4}$
- C $\frac{5\pi}{6}$
- D $\frac{3\pi}{4}$

Q:16

Topic Name:Mathematics-Section A

ItemCode:1916

The mean and variance of the data 4, 5, 6, 6, 7, 8, x , y , where $x < y$, are 6 and $\frac{9}{4}$

Question: respectively. Then $x^4 + y^2$ is equal to

- A 162
- B 320
- C 674
- D 420

Q:17

Topic Name:Mathematics-Section A

ItemCode:1917	
If a point $A(x, y)$ lies in the region bounded by the y -axis, straight lines	
Question: $2y + x = 6$ and $5x - 6y = 30$, then the probability that $y < 1$ is	
A	$\frac{1}{6}$
B	$\frac{5}{6}$
C	$\frac{2}{3}$
D	$\frac{6}{7}$

Q:18

Topic Name:Mathematics-Section A

ItemCode:1918	
The value of $\cot\left(\sum_{n=1}^{50} \tan^{-1}\left(\frac{1}{1+n+n^2}\right)\right)$ is	
Question:	
A	$\frac{26}{25}$
B	$\frac{25}{26}$
C	$\frac{50}{51}$
D	$\frac{52}{51}$

Q:19

Topic Name:Mathematics-Section A

ItemCode:1919	
Question: $\alpha = \sin 36^\circ$ is a root of which of the following equation?	
A	$16x^4 - 10x^2 - 5 = 0$
B	$16x^4 + 20x^2 - 5 = 0$
C	$16x^4 - 20x^2 + 5 = 0$
D	$16x^4 - 10x^2 + 5 = 0$

Q:20

Topic Name:Mathematics-Section A

ItemCode:1920	
Question:Which of the following statement is a tautology ?	
A	$((\sim q) \wedge p) \wedge q$
B	$((\sim q) \wedge p) \wedge (p \wedge (\sim p))$
C	$((\sim q) \wedge p) \vee (p \vee (\sim p))$
D	$(p \wedge q) \wedge (\sim(p \wedge q))$

Q:21

Topic Name:Mathematics-Section B

ItemCode:1921

Let $S = \{1, 2, 3, 4, 5, 6, 7, 8, 9, 10\}$. Define $f : S \rightarrow S$ as

$$f(n) = \begin{cases} 2n & , \text{ if } n = 1, 2, 3, 4, 5 \\ 2n - 11 & , \text{ if } n = 6, 7, 8, 9, 10 \end{cases}$$

Let $g : S \rightarrow S$ be a function such that $fog(n) = \begin{cases} n+1 & , \text{ if } n \text{ is odd} \\ n-1 & , \text{ if } n \text{ is even} \end{cases}$,

Question: Then $g(10)(g(1) + g(2) + g(3) + g(4) + g(5))$ is equal to _____.

Q:22

Topic Name:Mathematics-Section B

ItemCode:1922

Let α, β be the roots of the equation $x^2 - 4\lambda x + 5 = 0$ and α, γ be the roots of the equation $x^2 - (3\sqrt{2} + 2\sqrt{3})x + 7 + 3\lambda\sqrt{3} = 0$, $\lambda > 0$. If $\beta + \gamma = 3\sqrt{2}$, then

Question: $(\alpha + 2\beta + \gamma)^2$ is equal to _____.

Q:23

Topic Name:Mathematics-Section B

ItemCode:1923

Let A be a matrix of order 2×2 , whose entries are from the set $\{0, 1, 2, 3, 4, 5\}$. If the sum of all the entries of A is a prime number p , $2 < p < 8$, then the number of

Question: such matrices A is _____.

Q:24

Topic Name:Mathematics-Section B

ItemCode:1924

If the sum of the coefficients of all the positive powers of x , in the Binomial

expansion of $\left(x^n + \frac{2}{x^5}\right)^7$ is 939, then the sum of all the possible integral values

Question: of n is _____.

Q:25

Topic Name:Mathematics-Section B

ItemCode:1925

Let $[t]$ denote the greatest integer $\leq t$ and $\{t\}$ denote the fractional part of t . The integral value of α for which the left hand limit of the function

$f(x) = [1+x] + \frac{\alpha^{2[x]+\{x\}} + [x] - 1}{2[x] + \{x\}}$ at $x = 0$ is equal to $\alpha - \frac{4}{3}$, is _____.

Question:

Q:26

Topic Name:Mathematics-Section B

ItemCode:1926

If $y(x) = (x^x)^x$, $x > 0$, then $\frac{d^2x}{dy^2} + 20$ at $x = 1$ is equal to _____.

Question:

Q:27

Topic Name:Mathematics-Section B

ItemCode:1927

If the area of the region $\left\{(x, y) : x^{\frac{2}{3}} + y^{\frac{2}{3}} \leq 1, x + y \geq 0, y \geq 0\right\}$ is A , then $\frac{256A}{\pi}$ is

Question: equal to _____.

Q:28

Topic Name:Mathematics-Section B

ItemCode:1928

Let $y = y(x)$ be the solution of the differential equation

$$(1-x^2)dy = (xy + (x^3 + 2)\sqrt{1-x^2})dx, -1 < x < 1, \text{ and } y(0) = 0. \text{ If}$$

$$\int_{-\frac{1}{2}}^{\frac{1}{2}} \sqrt{1-x^2} y(x) dx = k, \text{ then } k^{-1} \text{ is equal to } \underline{\hspace{2cm}}.$$

Question:

Q:29

Topic Name:Mathematics-Section B

ItemCode:1929

Let a circle C of radius 5 lie below the x -axis. The line $L_1 : 4x + 3y + 2 = 0$ passes through the centre P of the circle C and intersects the line $L_2 : 3x - 4y - 11 = 0$ at Q . The line L_2 touches C at the point Q . Then the distance of P from the line

Question: $5x - 12y + 51 = 0$ is _____.

Q:30

Topic Name:Mathematics-Section B

ItemCode:1930

Let $S = \{E_1, E_2, \dots, E_8\}$ be a sample space of a random experiment such that

$$P(E_n) = \frac{n}{36} \text{ for every } n = 1, 2, \dots, 8. \text{ Then the number of elements in the set}$$

Question: $\left\{A \subseteq S : P(A) \geq \frac{4}{5}\right\}$ is _____.

Q:31

Topic Name:Physics-Section A

ItemCode:1931

The SI unit of a physical quantity is pascal-second. The dimensional formula of

Question: this quantity will be :

A $[ML^{-1}T^{-1}]$

B $[ML^{-1}T^{-2}]$

C $[ML^2T^{-1}]$

D $[M^{-1}L^3T^0]$

Q:32

Topic Name:Physics-Section A

ItemCode:1932

The distance of the Sun from earth is 1.5×10^{11} m and its angular diameter is

Question: (2000) s when observed from the earth. The diameter of the Sun will be :

A 2.45×10^{10} m

B 1.45×10^{10} m

C 1.45×10^9 m

D 0.14×10^9 m

Q:33

Topic Name:Physics-Section A

ItemCode:1933

When a ball is dropped into a lake from a height 4.9 m above the water level, it hits the water with a velocity v and then sinks to the bottom with the constant velocity v . It reaches the bottom of the lake 4.0 s after it is dropped. The

Question: approximate depth of the lake is :

A 19.6 m

B 29.4 m

C 39.2 m

D 73.5 m

Q:34

Topic Name:Physics-Section A

ItemCode:1934

One end of a massless spring of spring constant k and natural length l_0 is fixed while the other end is connected to a small object of mass m lying on a frictionless table. The spring remains horizontal on the table. If the object is made to rotate at an angular velocity ω about an axis passing through fixed end, then the elongation of the spring will be :

A $\frac{k - m\omega^2 l_0}{m\omega^2}$

B $\frac{m\omega^2 l_0}{k + m\omega^2}$

C $\frac{m\omega^2 l_0}{k - m\omega^2}$

D $\frac{k + m\omega^2 l_0}{m\omega^2}$

Q:35

Topic Name:Physics-Section A

ItemCode:1935

A stone tied to a string of length L is whirled in a vertical circle with the other end of the string at the centre. At a certain instant of time, the stone is at its lowest position and has a speed u . The magnitude of change in its velocity, as it reaches a position where the string is horizontal, is $\sqrt{x(u^2 - gL)}$. The value of x is-

A 3

B 2

C 1

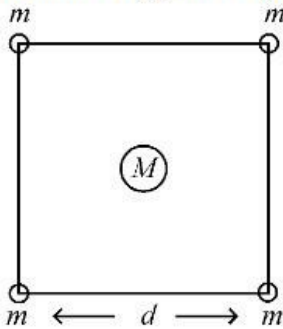
D 5

Q:36

Topic Name:Physics-Section A

ItemCode:1936

Four spheres each of mass m form a square of side d (as shown in figure). A fifth sphere of mass M is situated at the centre of square. The total gravitational potential energy of the system is :



Question:

A $-\frac{Gm}{d}[(4 + \sqrt{2})m + 4\sqrt{2}M]$

B $-\frac{Gm}{d}[(4 + \sqrt{2})M + 4\sqrt{2}m]$

C $-\frac{Gm}{d}[3m^2 + 4\sqrt{2}M]$

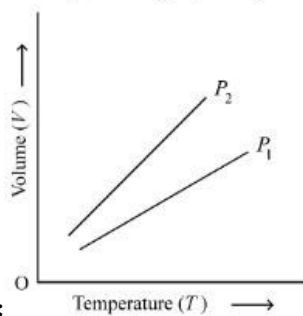
D $-\frac{Gm}{d}[6m^2 + 4\sqrt{2}M]$

Q:37

Topic Name:Physics-Section A

ItemCode:1937

For a perfect gas, two pressures P_1 and P_2 are shown in figure. The graph shows :



Question:

- | | |
|---|--|
| A | $P_1 > P_2$ |
| B | $P_1 < P_2$ |
| C | $P_1 = P_2$ |
| D | Insufficient data to draw any conclusion |

Q:38

Topic Name:Physics-Section A

ItemCode:1938

According to kinetic theory of gases,

A. The motion of the gas molecules freezes at 0°C .

B. The mean free path of gas molecules decreases if the density of molecules is increased.

C. The mean free path of gas molecules increases if temperature is increased keeping pressure constant.

D. Average kinetic energy per molecule per degree of freedom is $\frac{3}{2}k_B T$ (for monoatomic gases).

Question: Choose the most appropriate answer from the options given below:

- | | |
|---|--------------|
| A | A and C only |
| B | B and C only |
| C | A and B only |
| D | C and D only |

Q:39

Topic Name:Physics-Section A

ItemCode:1939

A lead bullet penetrates into a solid object and melts. Assuming that 40% of its kinetic energy is used to heat it, the initial speed of bullet is :

(Given, initial temperature of the bullet = 127°C ,

Melting point of the bullet = 327°C ,

Latent heat of fusion of lead = $2.5 \times 10^4 \text{ J kg}^{-1}$,

Question: Specific heat capacity of lead = 125 J/kg K)

- | | |
|---|-----------------------|
| A | 125 ms^{-1} |
| B | 500 ms^{-1} |
| C | 250 ms^{-1} |
| D | 600 ms^{-1} |

Q:40

Topic Name:Physics-Section A

ItemCode:1940

The equation of a particle executing simple harmonic motion is given by

$x = \sin \pi \left(t + \frac{1}{3} \right) \text{ m}$. At $t = 1 \text{ s}$, the speed of particle will be

Question:(Given: $\pi = 3.14$) -

- | | |
|---|-------------------------|
| A | 0 cm s^{-1} |
| B | 157 cm s^{-1} |

C 272 cm s^{-1}

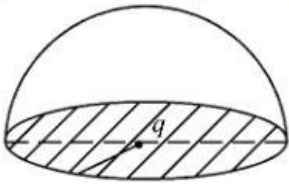
D 314 cm s^{-1}

Q:41

Topic Name:Physics-Section A

ItemCode:1941

If a charge q is placed at the centre of a closed hemispherical non-conducting surface, the total flux passing through the flat surface would be :



Question:

A $\frac{q}{\epsilon_0}$

B $\frac{q}{2\epsilon_0}$

C $\frac{q}{4\epsilon_0}$

D $\frac{q}{2\pi\epsilon_0}$

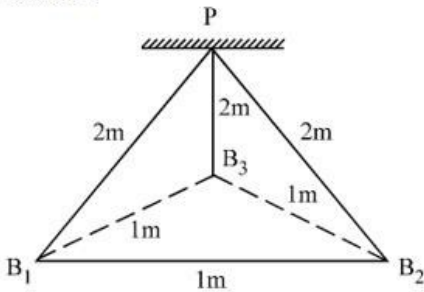
Q:42

Topic Name:Physics-Section A

ItemCode:1942

Three identical charged balls each of charge 2 C are suspended from a common point P by silk threads of 2 m each (as shown in figure). They form an equilateral triangle of side 1 m .

The ratio of net force on a charged ball to the force between any two charged balls will be :



Question:

A 1:1

B 1:4

C $\sqrt{3}:2$

D $\sqrt{3}:1$

Q:43

Topic Name:Physics-Section A

ItemCode:1943

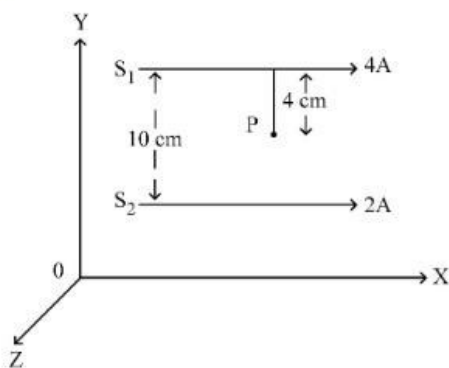
Two long parallel conductors S_1 and S_2 are separated by a distance 10 cm and carrying currents of 4A and 2A respectively. The conductors are placed along x – axis in X - Y plane. There is a point P located between the conductors (as shown in figure).

A charge particle of 3π coulomb is passing through the point P with velocity

$\vec{v} = (2\hat{i} + 3\hat{j})$ m/s; where $\hat{i}$ & $\hat{j}$ represents unit vector along x & y axis

respectively.

The force acting on the charge particle is $4\pi \times 10^{-5}(-x\hat{i} + 2\hat{j})$ N. The value of x is :



Question:

- A 2
- B 1
- C 3
- D -3

Q:44

Topic Name:Physics-Section A

ItemCode:1944

If L , C and R are the self inductance, capacitance and resistance respectively,

Question: which of the following does not have the dimension of time?

- A RC
- B $\frac{L}{R}$
- C $\sqrt{LC}$
- D $\frac{L}{C}$

Q:45

Topic Name:Physics-Section A

ItemCode:1945

Given below are two statements :

Statement I: A time varying electric field is a source of changing magnetic field and vice-versa. Thus a disturbance in electric or magnetic field creates EM waves.

Statement II: In a material medium, the EM wave travels with speed $v = \frac{1}{\sqrt{\mu_0 \epsilon_0}}$.

In the light of the above statements, choose the correct answer from the options

Question: given below.

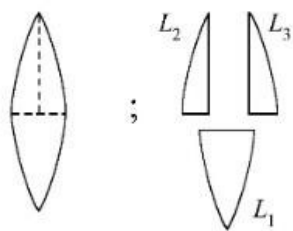
- A Both statement I and statement II are true
- B Both statement I and statement II are false
- C Statement I is correct but statement II is false
- D Statement I is incorrect but statement II is true

Q:46

Topic Name:Physics-Section A

ItemCode:1946

A convex lens has power P. It is cut into two halves along its principal axis. Further one piece (out of the two halves) is cut into two halves perpendicular to the principal axis (as shown in figures). Choose the incorrect option for the reported pieces.



Question:

- A Power of $L_1 = \frac{P}{2}$
- B Power of $L_2 = \frac{P}{2}$
- C Power of $L_3 = \frac{P}{2}$
- D Power of $L_1 = P$

Q:47

Topic Name:Physics-Section A

ItemCode:1947

Question: If a wave gets refracted into a denser medium, then which of the following is true?

- A wavelength, speed and frequency decreases.
- B wavelength increases, speed decreases and frequency remains constant.
- C wavelength and speed decreases but frequency remains constant.
- D wavelength, speed and frequency increases.

Q:48

Topic Name:Physics-Section A

ItemCode:1948

Given below are two statements:

Statement I: In hydrogen atom, the frequency of radiation emitted when an electron jumps from lower energy orbit (E_1) to higher energy orbit (E_2), is given as

$$hf = E_1 - E_2$$

Statement II: The jumping of electron from higher energy orbit (E_2) to lower energy orbit (E_1) is associated with frequency of radiation given as

$$f = (E_2 - E_1)/h$$

This condition is Bohr's frequency condition.

In the light of the above statements, choose the correct answer from the options

Question: given below:

- A Both statement I and statement II are true.
- B Both statement I and statement II are false.
- C Statement I is correct but statement II is false.
- D Statement I is incorrect but statement II is true.

Q:49

Topic Name:Physics-Section A

ItemCode:1949

Question: For a transistor to act as a switch, it must be operated in

- A Active region.
- B Saturation state only.
- C Cut-off state only.
- D Saturation and cut-off state.

Q:50

Topic Name:Physics-Section A

ItemCode:1950

We do not transmit low frequency signal to long distances because-

- (a) The size of the antenna should be comparable to signal wavelength which is unreal solution for a signal of longer wavelength.
- (b) Effective power radiated by a long wavelength baseband signal would be high.
- (c) We want to avoid mixing up signals transmitted by different transmitter simultaneously.
- (d) Low frequency signal can be sent to long distances by superimposing with a high frequency wave as well.

Question: Therefore, the most suitable option will be :

- A All statements are true
- B (a), (b) and (c) are true only
- C (a), (c) and (d) are true only
- D (b), (c) and (d) are true only

Q:51

Topic Name:Physics-Section B

ItemCode:1951

A mass of 10 kg is suspended vertically by a rope of length 5 m from the roof. A force of 30 N is applied at the middle point of rope in horizontal direction. The angle made by upper half of the rope with vertical is $\theta = \tan^{-1}(x \times 10^{-1})$. The value of x is _____.

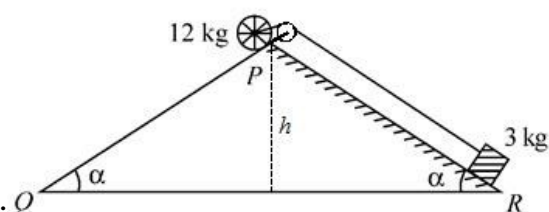
Question: (Given, $g = 10 \text{ m/s}^2$)

Q:52

Topic Name:Physics-Section B

ItemCode:1952

A rolling wheel of 12 kg is on an inclined plane at position P and connected to a mass of 3 kg through a string of fixed length and pulley as shown in figure. Consider PR as friction free surface. The velocity of centre of mass of the wheel when it reaches at the bottom Q of the inclined plane PQ will be $\frac{1}{2}\sqrt{xgh} \text{ m/s}$. The value of x is _____.



Question:

Q:53

Topic Name:Physics-Section B

ItemCode:1953

A diatomic gas ($\gamma = 1.4$) does 400J of work when it is expanded isobarically. The

Question: heat given to the gas in the process is _____ J.

Q:54

Topic Name:Physics-Section B

ItemCode:1954

A particle executes simple harmonic motion. Its amplitude is 8 cm and time period is 6 s. The time it will take to travel from its position of maximum displacement to

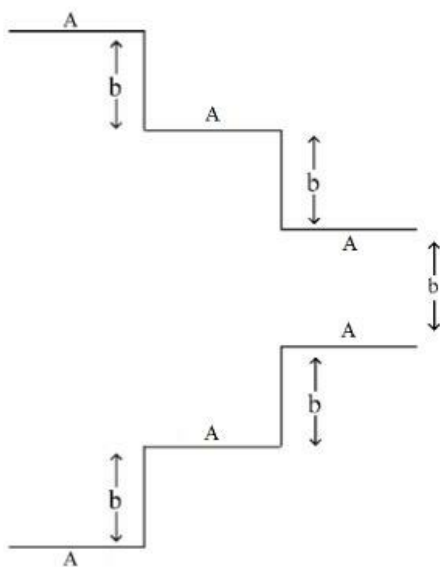
Question: the point corresponding to half of its amplitude, is _____ s.

Q:55

Topic Name:Physics-Section B

ItemCode:1955

A parallel plate capacitor is made up of stair like structure with a plate area A of each stair and that is connected with a wire of length b , as shown in the figure. The capacitance of the arrangement is $\frac{x}{15} \frac{\epsilon_0 A}{b}$, The value of x is _____.



Question:

Q:56

Topic Name:Physics-Section B

ItemCode:1956

The current density in a cylindrical wire of radius $r = 4.0 \text{ mm}$ is $1.0 \times 10^6 \text{ A/m}^2$.

The current through the outer portion of the wire between radial distances $\frac{r}{2}$ and r

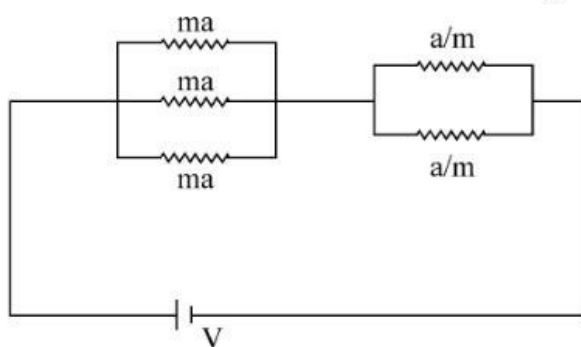
Question: is $x\pi A$; where x is _____.

Q:57

Topic Name:Physics-Section B

ItemCode:1957

In the given circuit 'a' is an arbitrary constant. The value of m for which the equivalent circuit resistance is minimum, will be $\sqrt{\frac{x}{2}}$. The value of x is _____.



Question:

Q:58

Topic Name:Physics-Section B

ItemCode:1958

A deuteron and a proton moving with equal kinetic energy enter into to a uniform magnetic field at right angle to the field. If r_d and r_p are the radii of their circular

paths respectively, then the ratio $\frac{r_d}{r_p}$ will be $\sqrt{x} : 1$ where x is _____.

Question:

Q:59

Topic Name:Physics-Section B

ItemCode:1959

A metallic rod of length 20 cm is placed in North-South direction and is moved at a constant speed of 20 m/s towards East. The horizontal component of the Earth's magnetic field at that place is 4×10^{-3} T and the angle of dip is 45° . The emf

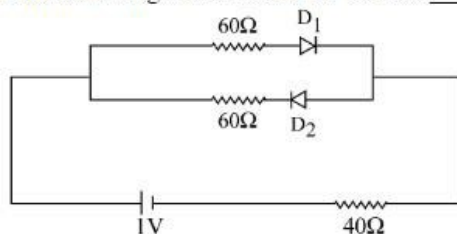
Question: induced in the rod is _____ mV.

Q:60

Topic Name:Physics-Section B

ItemCode:1960

The cut-off voltage of the diodes (shown in figure) in forward bias is 0.6 V. The current through the resistor of 40Ω is _____ mA.



Question:

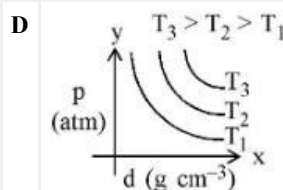
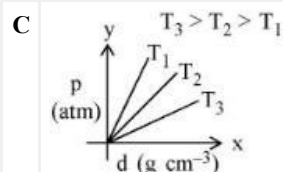
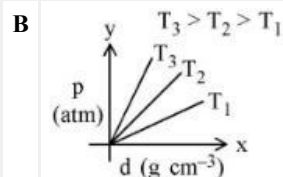
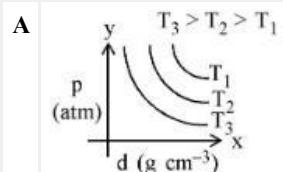
Q:61

Topic Name:Chemistry-Section A

ItemCode:1961

Which amongst the given plots is the correct plot for pressure (p) vs density (d) for an ideal gas?

Question:



Q:62

Topic Name:Chemistry-Section A

ItemCode:1962

Question: Identify the **incorrect** statement for PCl_5 from the following.

- A In this molecule, orbitals of phosphorous are assumed to undergo sp^3d hybridization.
- B The geometry of PCl_5 is trigonal bipyramidal.
- C PCl_5 has two axial bonds stronger than three equatorial bonds.
- D The three equatorial bonds of PCl_5 lie in a plane.

Q:63

Topic Name:Chemistry-Section A

ItemCode:1963

Statement I: Leaching of gold with cyanide ion in absence of air / O_2 leads to cyano complex of Au(III).

Statement II: Zinc is oxidized during the displacement reaction carried out for gold extraction.

In the light of the above statements, choose the **correct** answer from the options

Question: given below.

- | | |
|---|---|
| A | Both statement I and statement II are correct. |
| B | Both statement I and statement II are incorrect. |
| C | Statement I is correct but statement II is incorrect. |
| D | Statement I is incorrect but statement II is correct. |

Q:64

Topic Name:Chemistry-Section A

ItemCode:1964

Question: The correct order of increasing intermolecular hydrogen bond strength is

- | | |
|---|---------------------|
| A | $HCN < H_2O < NH_3$ |
| B | $HCN < CH_4 < NH_3$ |
| C | $CH_4 < HCN < NH_3$ |
| D | $CH_4 < NH_3 < HCN$ |

Q:65

Topic Name:Chemistry-Section A

ItemCode:1965

Question: The correct order of increasing ionic radii is

- | | |
|---|--|
| A | $Mg^{2+} < Na^+ < F^- < O^{2-} < N^{3-}$ |
| B | $N^{3-} < O^{2-} < F^- < Na^+ < Mg^{2+}$ |
| C | $F^- < Na^+ < O^{2-} < Mg^{2+} < N^{3-}$ |
| D | $Na^+ < F^- < Mg^{2+} < O^{2-} < N^{3-}$ |

Q:66

Topic Name:Chemistry-Section A

ItemCode:1966

The gas produced by treating an aqueous solution of ammonium chloride with

Question: sodium nitrite is

- | | |
|---|--------|
| A | NH_3 |
| B | N_2 |
| C | N_2O |
| D | Cl_2 |

Q:67

Topic Name:Chemistry-Section A

ItemCode:1967

Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**.

Assertion A: Flourine forms one oxoacid.

Reason R: Flourine has smallest size amongst all halogens and is highly electronegative.

In the light of the above statements, choose the *most appropriate* answer from the

Question: options given below.

- | | |
|---|--|
| A | Both A and R are correct and R is the correct explanation of A . |
| B | Both A and R are correct but R is NOT the correct explanation of A . |
| C | A is correct but R is not correct. |

D A is not correct but R is correct.

Q:68

Topic Name:Chemistry-Section A

ItemCode:1968

Question: In 3d series, the metal having the highest M^{2+}/M standard electrode potential is

- | | |
|---|----|
| A | Cr |
| B | Fe |
| C | Cu |
| D | Zn |

Q:69

Topic Name:Chemistry-Section A

ItemCode:1969

The 'f' orbitals are half and completely filled, respectively in lanthanide ions

Question:[Given : Atomic no. Eu, 63; Sm, 62; Tm, 69; Tb, 65; Yb, 70; Dy, 66]

- | | |
|---|-------------------------|
| A | Eu^{2+} and Tm^{2+} |
| B | Sm^{2+} and Tm^{3+} |
| C | Tb^{4+} and Yb^{2+} |
| D | Dy^{3+} and Yb^{3+} |

Q:70

Topic Name:Chemistry-Section A

ItemCode:1970

Arrange the following coordination compounds in the increasing order of magnetic moments. (Atomic numbers: Mn = 25; Fe = 26)

- A. $[FeF_6]^{3-}$
B. $[Fe(CN)_6]^{3-}$
C. $[MnCl_6]^{3-}$ (high spin)
D. $[Mn(CN)_6]^{3-}$

Question: Choose the correct answer from the options given below:

- | | |
|---|-----------------|
| A | $A < B < D < C$ |
| B | $B < D < C < A$ |
| C | $A < C < D < B$ |
| D | $B < D < A < C$ |

Q:71

Topic Name:Chemistry-Section A

ItemCode:1971

On the surface of polar stratospheric clouds, hydrolysis of chlorine nitrate gives A and B while its reaction with HCl produces B and C. A, B and C are, respectively

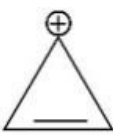
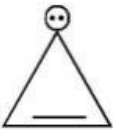
- | | |
|---|---------------------------|
| A | HOCl, HNO_3 , Cl_2 |
| B | Cl_2 , HNO_3 , HOCl |
| C | $HClO_2$, HNO_2 , HOCl |
| D | HOCl, HNO_2 , Cl_2O |

Q:72

Topic Name:Chemistry-Section A

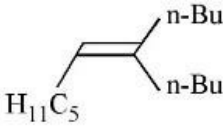
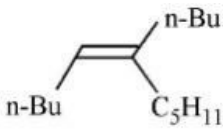
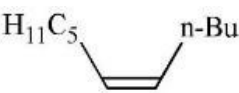
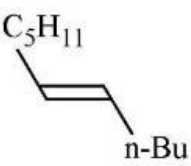
ItemCode:1972

Question: Which of the following is most stable?

A	
B	
C	
D	

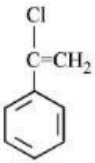
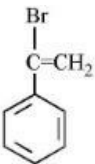
Q:73

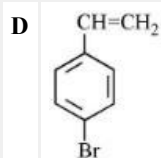
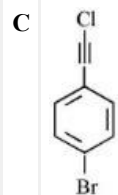
Topic Name: Chemistry-Section A

ItemCode:1973 What will be the major product of following sequence of reactions? $\text{n-Bu} \equiv \xrightarrow[\text{(ii) Lindlar cat, H}_2]{\text{(i) n-BuLi, n-C}_5\text{H}_{11}\text{Cl}}$	
Question:	
A	
B	
C	
D	

Q:74

Topic Name: Chemistry-Section A

ItemCode:1974 Product 'A' of following sequence of reactions is $\text{Ethylbenzene} \xrightarrow[\text{(c) alc. KOH}]{\text{(a) Br}_2, \text{Fe} \quad \text{(b) Cl}_2, \Delta} \text{'A'}$ Major Product	
Question:	
A	
B	



Q:75
Topic Name:Chemistry-Section A

ItemCode:1975
 Match **List I** with **List II**.

List I		List II
A.		I. Br ₂ in CS ₂
B.		II. Na ₂ Cr ₂ O ₇ /H ₂ SO ₄
C.		III. Zn
D.		IV. CHCl ₃ /NaOH

Question: Choose the correct answer from the options given below:

- A** A-IV, B-III, C-II, D-I
B A-IV, B-III, C-I, D-II
C A-II, B-III, C-I, D-IV
D A-IV, B-II, C-III, D-I

Q:76
Topic Name:Chemistry-Section A

ItemCode:1976
 Decarboxylation of all six possible forms of diaminobenzoic acids
 C₆H₃(NH₂)₂COOH yields three products A, B and C. Three acids give a product 'A', two acids give a product 'B' and one acid give a product 'C'. The melting

Question: point of product 'C' is

- A** 63°C
B 90°C
C 104°C
D 142°C

Q:77
Topic Name:Chemistry-Section A

ItemCode:1977
Question: Which is true about Buna-N?

- A** It is a linear polymer of 1,3-butadiene.
B It is obtained by copolymerization of 1,3-butadiene and styrene.

C It is obtained by copolymerization of 1,3-butadiene and acrylonitrile.

D The suffix N in Buna-N stands for its natural occurrence.

Q:78

Topic Name:Chemistry-Section A

ItemCode:1978

Given below are two statements.

Statement I: Maltose has two α -D-glucose units linked at C_1 and C_4 and is a reducing sugar.

Statement II: Maltose has two monosaccharides: α -D-glucose and β -D-glucose linked at C_1 and C_6 and it is a non-reducing sugar.

In the light of the above statements, choose the **correct** answer from the options

Question: given below.

A Both Statement I and Statement II are true.

B Both Statement I and Statement II are false.

C Statement I is true but Statement II is false.

D Statement I is false but Statement II is true.

Q:79

Topic Name:Chemistry-Section A

ItemCode:1979

Match List I with List II.

List I	List II
A. Antipyretic	I. Reduces pain
B. Analgesic	II. Reduces stress
C. Tranquilizer	III. Reduces fever
D. Antacid	IV. Reduces acidity(stomach)

Question: Choose the correct answer from the options given below:

A A-III, B-I, C-II, D-IV

B A-III, B-I, C-IV, D-II

C A-I, B-IV, C-II, D-III

D A-I, B-III, C-II, D-IV

Q:80

Topic Name:Chemistry-Section A

ItemCode:1980

Match List I with List II.

List I (Anion)	List II (gas evolved on reaction with $\text{dil.H}_2\text{SO}_4$)
A. CO_3^{2-}	I. Colourless gas which turns lead acetate paper black.
B. S^{2-}	II. Colourless gas which turns acidified potassium dichromate solution green.
C. SO_3^{2-}	III. Brown fumes which turns acidified KI solution containing starch blue.
D. NO_2^-	IV. Colourless gas evolved with brisk effervescence, which turns lime water milky.

Question: Choose the correct answer from the options given below:

A A-III, B-I, C-II, D-IV

B	A-II, B-I, C-IV, D-III
C	A-IV, B-I, C-III, D-II
D	A-IV, B-I, C-II, D-III

Q:81
 Topic Name:Chemistry-Section B

ItemCode:1981
 116 g of a substance upon dissociation reaction, yields 7.5 g of hydrogen, 60 g of oxygen and 48.5 g of carbon. Given that the atomic masses of H, O and C are 1, 16 and 12, respectively. The data agrees with how many formulae of the following ?
 Question: _____.
 A. CH_3COOH B. HCHO C. CH_3OOCH_3 D. CH_3CHO

Q:82
 Topic Name:Chemistry-Section B

ItemCode:1982
 Consider the following set of quantum numbers.

	n	l	m_l
A.	3	3	-3
B.	3	2	-2
C.	2	1	+1
D.	2	2	+2

 Question: The number of correct sets of quantum numbers is _____.

Q:83
 Topic Name:Chemistry-Section B

ItemCode:1983
 BeO reacts with HF in presence of ammonia to give [A] which on thermal decomposition produces [B] and ammonium fluoride. Oxidation state of Be in [A] is _____
 Question: is _____

Q:84
 Topic Name:Chemistry-Section B

ItemCode:1984
 When 5 moles of He gas expand isothermally and reversibly at 300 K from 10 litre to 20 litre, the magnitude of the maximum work obtained is _____ J. [nearest integer](Given : $R = 8.3 \text{ J K}^{-1} \text{ mol}^{-1}$ and $\log 2 = 0.3010$)
 Question: _____

Q:85
 Topic Name:Chemistry-Section B

ItemCode:1985
 A solution containing $2.5 \times 10^{-3} \text{ kg}$ of a solute dissolved in $75 \times 10^{-3} \text{ kg}$ of water boils at 373.535 K. The molar mass of the solute is _____ g mol^{-1} . [nearest integer] (Given : $K_b(\text{H}_2\text{O}) = 0.52 \text{ K kg mol}^{-1}$ and boiling point of water = 373.15 K)
 Question: _____

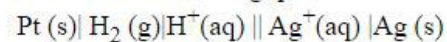
Q:86
 Topic Name:Chemistry-Section B

ItemCode:1986
 Question: pH value of 0.001 M NaOH solution is _____.

Q:87
 Topic Name:Chemistry-Section B

ItemCode:1987

For the reaction taking place in the cell:



$$E^\circ_{\text{cell}} = +0.5332 \text{ V.}$$

Question: The value of $\Delta_f G^\ominus$ is _____ kJ mol^{-1} . (in nearest integer)

Q:88

Topic Name:Chemistry-Section B

ItemCode:1988

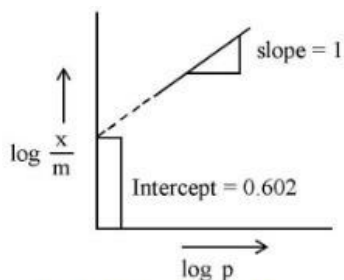
It has been found that for a chemical reaction with rise in temperature by 9 K the rate constant gets doubled. Assuming a reaction to be occurring at 300 K, the value of activation energy is found to be _____ kJ mol^{-1} . [nearest integer]

Question: (Given $\ln 10 = 2.3$, $R = 8.3 \text{ J K}^{-1} \text{ mol}^{-1}$, $\log 2 = 0.30$)

Q:89

Topic Name:Chemistry-Section B

ItemCode:1989



If the initial pressure of a gas is 0.03 atm, the mass of the gas adsorbed per gram of

Question: the adsorbent is _____ $\times 10^{-2} \text{ g}$.

Q:90

Topic Name:Chemistry-Section B

ItemCode:1990

0.25 g of an organic compound containing chlorine gave 0.40 g of silver chloride in Carius estimation. The percentage of chlorine present in the compound is _____. [in nearest integer]

Question: (Given: Molar mass of Ag is 108 g mol^{-1} and that of Cl is 35.5 g mol^{-1})